Electrophilic Reactions of ClCCo₃(CO)₉. A New Preparation of Arylmethinyltricobalt Enneacarbonyls

By R. DOLBY and B. H. ROBINSON*

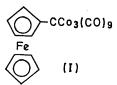
(Department of Chemistry, University of Otago, Dunedin, New Zealand)

Summary A convenient new preparation of arylmethinyltricobalt enneacarbonyls, involving the reaction of $ClCCo_3(CO)_9$ with the arene in the presence of a typical Friedel-Crafts reagent like AlCl₃, is described.

RECENTLY we investigated the reactions of methinyltricobalt enneacarbonyls, YCCo₃(CO)₉, with arenes and showed that while arene complexes $YCCo_3(CO)_6(arene)$ can be prepared when Y = Me, Ph, and F, the chloro- and bromoclusters yield the cobalt carbonyl carbides $Co_5(CO)_{15}C_3H$, $Co_{6}(CO)_{18}C_{4}$, and $Co_{8}(CO)_{24}C_{6}$.¹ In an attempt to prepare Cl(or Br)CCo₃(CO)₆(arene) derivatives, we tried AlCl₃ as catalyst, a method used successfully by Fischer and his co-workers.² However, instead of an arene complex forming, a ready Friedel-Crafts reaction occurred to give arylmethinyltricobalt enneacarbonyls in up to 80% yield. (Other Friedel-Crafts reagents such as FeCl₃ can be used).

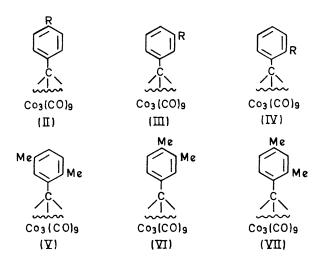
$$\operatorname{ClCCo}_3(\operatorname{CO})_{\mathfrak{g}} + \operatorname{RH} \xrightarrow{\operatorname{AlCl}_{\mathfrak{g}}} \operatorname{RCCo}_3(\operatorname{CO})_{\mathfrak{g}} + \operatorname{HCl}.$$

This is a very convenient route to the aryls, as ClCCo₂- $(CO)_{9}$ is very easily prepared from $Co_{2}(CO)_{8}$, whereas the other routes to the aryls are limited (where the preparation of $Co_2(CO)_6 RC \equiv CH$ compounds are involved³) or impossible if the 1,1,1-trichloroaryl is required.⁴ This method can be extended to solid arenes using dichloromethane as a solvent and a number of new aryl derivatives have been characterized including $R = ClC_{6}H_{4}$, $BrC_{6}H_{4}$, biphenyl, and o-, m-, and $p-C_{\theta}H_{3}(CH_{3})_{2}$ and the novel ferrocene derivative $(\pi - C_5 H_5) Fe[\pi - C_5 H_4 CCo_3 (CO)_9]$ of probable structure **(**I**)**.



Isomeric derivatives (II), (III), and (IV) could conceivably form in these Friedel-Crafts reactions, and n.m.r., i.r., and mass spectra have been used to elucidate the stereochemistries of the product isolated. In some cases isomers have been separated by gas chromatography using a 6 ft. 10% Carbowax 20M column. The spectral features that are thought to characterize compounds of structures (IV), (V) and (VII) are (i) common fragments in the mass spectra are migration fragments of the type Co₃R⁺, Co₂R⁺, CoR⁺, and Co₃CR⁺ which are absent in the spectra of para- and meta-substituted derivatives, the comparable migration fragments of the latter spectra being $Co_nH^+(n=3-1)$ and

 $Co_n CH^+$ (n=3-1); appropriate metastable transitions are observed. (ii) $YCCo_3(CO)_9$ derivatives of idealised C_{3n} symmetry, and those where there is no steric interaction between Y and the cluster, exhibit a typical i.r. spectrum in the carbonyl stretching region of four bands ca. 2100, 2050, 2040, and 2020 cm⁻¹, although five bands are predicted.⁵ However, in those cases where there are gross deviations from C_{3v} symmetry (for example the ferrocene derivative) or where there is steric interaction [structures (IV), (V), and (VII)], the band at 2050 cm^{-1} (two accidentally degenerate E modes⁵) becomes very broad and splits into two components.



With these criteria as a guide and using the ¹H n.m.r. spectra to confirm our structural assignments, it was found that iodobenzene, bromobenzene, and most alkylbenzenes are substituted in the *para*-position whereas chlorobenzene undergoes ortho-substitution. Only one structure is possible for the p-xylene derivative (V) but o- and m-xylene were substituted in the para- and ortho-position, respectively, (VI) and (VII). meta-Substitution was never observed, and strongly deactivated arenes e.g. nitrobenzene did not undergo these Friedel-Crafts reactions. Steric requirements are also important, as shown by the failure of mesitylene to react and the slow reactions and poor yields with m- and p-xylene and ferrocene.

The chloro-cluster also reacts with typical Grignard and lithium reagents and thus a wide variety of alkyl, alkene, and arvl substituted cluster derivatives can be prepared using these reagents and Friedel-Crafts reactions.

(Received, June 16th, 1970; Com. 935.)

- ¹ B. H. Robinson, J. L. Spencer, and R. Hodges, Chem. Comm., 1968, 1480.
 ² E. O. Fischer and W. Hafer, Z. Naturforsch., 1955, 10b, 665.
 ³ R. Markby, I. Wender, R. A. Friedel, F. A. Cotton, and H. W. Sternberg, J. Amer. Chem. Soc., 1958, 80, 6529.
 ⁴ W. T. Dent, L. A. Duncason, R. G. Guy, H. W. B. Reed, and B. L. Shaw, Proc. Chem. Soc., 1961, 169; G. Bor, B. Marko, and L. Marko, Chem. Ber., 1962, 95, 333; R. Ercoli, E. Santambrogio, and G. T. Casagrade, Chimica e Industria, 1962, 44, 1344.
 ⁵ G. Bor, Proceedings, Symposium on Co-ordination Chemistry, Tihany, Hungary, 1964, p. 361.